\$/180/62/006/005/005/011 E073/E535

Karelin, V.V., Nesmeyanov, An.N., Priselkov, Yu.A. (Moscow) AUTHURS:

TITLE: hore accurate data on the vapour pressure of metallic

PERFODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye tekhnicheskikh nauk. Metallurgiya i toplivo, no.5, 1962,

117-118

TEXT: In an earlier published paper, the authors studied the vapour pressure of metallic yttrium of a purity of 99.9% (without taking into consideration gaseous admixtures). According to those results, the vapour pressure can be expressed by

$$\log P_{\text{inm Hg}} = 7.8130 - \frac{15.803}{.T}$$
 (1)

This agreed with results obtained for yttrium of 99.5% purity (0.1% Ta, 0.4% 0₂). New investigations were carried out with high-purity yttrium containing only traces of metallic admixtures and less than 0.1% gaseous admixtures. The obtained data are tabulated for the temperature range 1132 to 1460°C. According Card 1/2

More accurate data on the vapour ... S/180/62/000/005/005/011 E073/E535

to these data, applying the method of least squares, the vapour pressure of high-purity metallic yttrium obeys the following

 $l_g P_{mm Hg} = 3.6786 - \frac{18}{T}$

From this, the sublimation heat was determined at 84.71 kcal/g-atom. The divergence between the here obtained and the earlier results is explained by the evaporation of volatile sub-oxides of yttrium, the existence of which was confirmed by means of a resonance mass spectrometer. A similar phenomenon was observed by Goldstein, Walsh and White (On the use of tantalum Knudsen cells in high temperature thermodynamic studies of oxides, J. Phys. Chem., 1960, 64, No.8, p.1087) who proved by means of a mass spectrometer that the increased rate of evaporation of La oxide from tantalum crucibles is caused by the reaction

 $3\text{Ta} + 4(1)\text{La}_2\text{O}_3 > 2\text{TaO} + \text{TaO}_2 + 8(1)\text{LaO}$.

The relative limit error in measuring the vapour pressure was ± 20 % for the radioactive and ± 24 % for the non-radioactive specimens. SUBMITTED:

Jun: 5, 1962

Card 2/2

KARELIN, V.V.; NESMEYANOV, A.N.; PRISEIKOV, Yu.A.; CHZHOU KUN'-IN

[Chou K'un-ying]

Measuring the vapor pressure of metallic yttrium. Vest.Mosk.un.

Ser.2: Khim. 17 no.2:40-41 Mr-Ap '62. (MIRA 15:4)

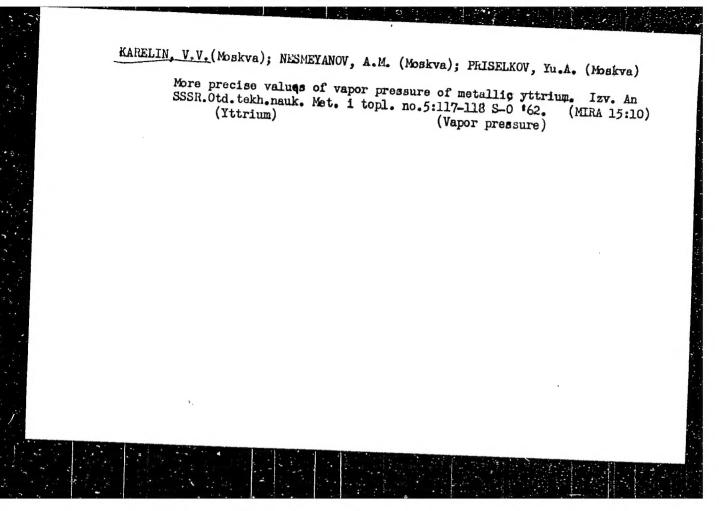
1. Kafedra radiokhimii Moskovskogo universiteta.

(Yttrium) (Vapor pressure)

MEN'KOV, A.A.; KOMISSAROVA, L.N.; KARELIN, V.V.; PRISELKOV, Yu.A.; NESMEYANOV, An.N.; SPITSYN, Vikt.I., akademik

Investigation of high-purity metallic scandium. Dokl.AN SSSR 144 no.1:122-125 My 162. (MIRA 15:5)

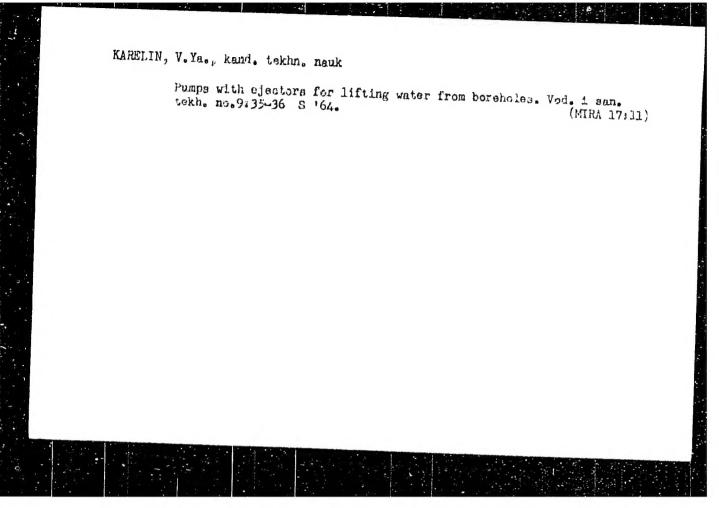
1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova. (Scandium)



KARELIN, V.V.; NESMEYANOV, An.N.; PRISELKOV, Yu.A.

Vapor pressure of metallic scandium. Dokl.AN SSSR 144 no.2:352-354 My 162. (MIRA 15:5)

l. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova.
Predstavleno akademikom Vikt.I.Spitsynym.
(Scandium) (Vapor pressure)



APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

GUBIN, M.F., dots., kand.tekhn.nauk; KARMLIN, V.Ya., inzh.

Effect of varying pressure of model turbines on their characteristics. Nauch.dokl.vys.shkoly; stroi. no.2:259-263 '58.

(Hydraulic turbines--Models)

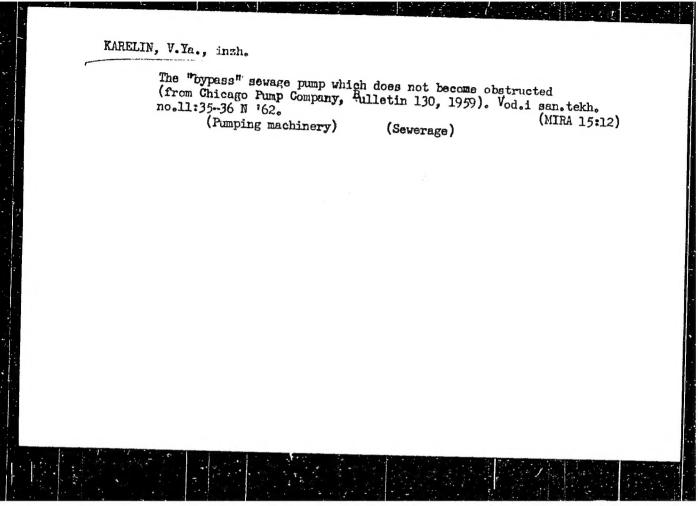
(Hydraulic turbines--Models)

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

KAPELIN, V. Ya. Cand Tech bei -- (diss) "Effect of bent suction tutes upon the operation of propeller and rotary-vens hydroturbines." Mos, 1958

20 pp; 4 sheets of diagrams (Min of Higher Education USDR. Mos Order of Labor Red Benner Construction Engineering Inst im V. V. Kuybyshev), 150 copies (KL, 52-58, 102)

-52-



S/285/63/000/002/012/012 A052/A126

AUTHOR:

Karelin, V.Ya.

TITLE:

The effect of hydrodynamic conditions and liquid temperature on the beginning and development of cavitation

PERIODICAL: Referativnyy zhurnal. Otdel'nyy vypusk. 49. Turbostroyeniye, no. 2, 1963, 25, abstract 2.49.134. (Sb. tr. Mosk. inzh. -stroit. in-t, no. 40, 1962, 49 - 55)

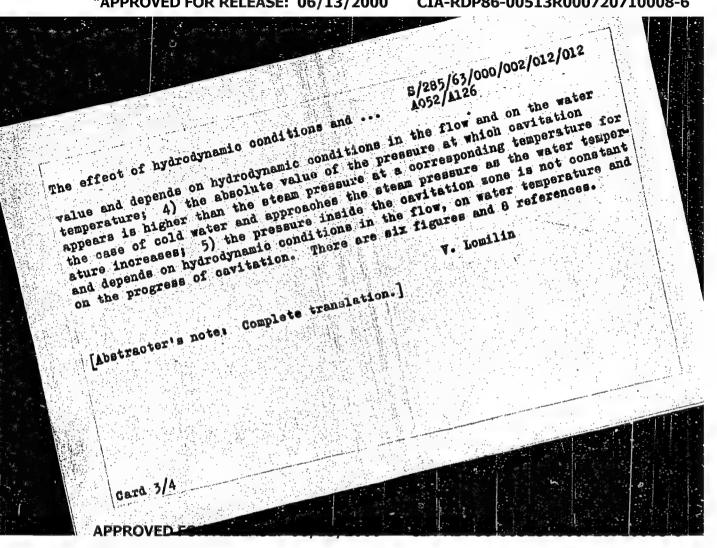
TEXT: For the experiments a Venturi flowmeter 4 was used with a cylindrical insert in the oblate section (see the Figure); the flowmeter 4 was installed between the flowmeter 5 and the pumping unit 3-3 supplying water from the water line through the steam-to-water preheater 1 and water heater with a deaerator 2. As the experiments have shown at any time constant water temperature, an increase of pressure in the input of the flow-mater inhibits the beginning of cavitation and increases the maximum flow. At a constant pressure in the input an increase of water temperature accelent forms of cavitation were observed during the experiments: initial Card 1/4

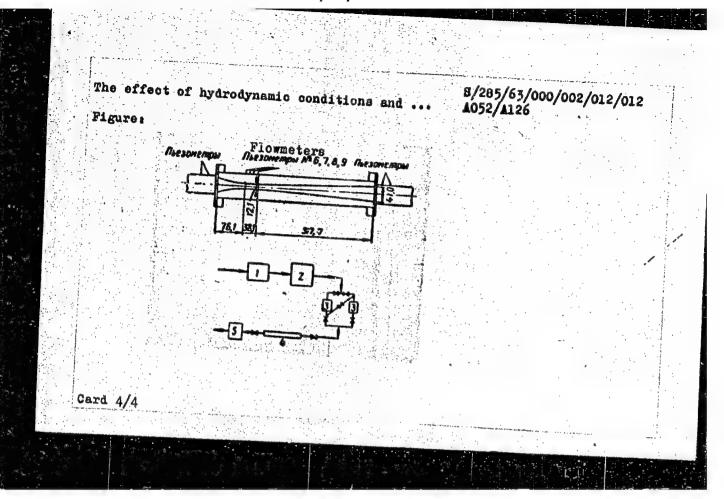
The effect of hydrodynamic conditions and ..

B/285/63/000/002/012/012 A052/A126

cavitation appeared in the form of a small steady cavern directly at the orifice of the piezometer 7. This form of cavitation is dissimilar from that which is typical for water-measuring devices with a contraction. It had no noticeable effect on the flow characteristic of the flowmeter and its early appearance was explained by a local roughness of the surface caused by sharp edges of the piezometric orifice. With a further increase of the flow or a decraese of the general pressure level in the system an unsteady annular cavern appeared in the flowmeter section near the piezometric orifice 9 which later stabilized and assumed a characteristic form. The moment of appearance of the unsteady annular cavern was considered as the beginning of cavitation in the orifice 9 or in the flowmeter in general The results of the investigation have shown that 1) at equal values hin characterizing the pressure in the input of the flowmeter and corresponding to the excess pressure in the suction pipe in the case of pumps, the flow values necessary for the beginning of cavitation do not depend on temperature; 2) the development of cavitation in the flowmeter takes an extraordinarily rapid course on account of which the flow value necessary for the beginning of cavitation and the capacity of the flowmeter are close to one another; 3) the pressure at which cavitation appears is not a constant

Card 2/4





APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

L 16726-63 EPA(b)/EWT(1)/EWP(q)/EWT(m)/BDS

AEDC/AFFTC/ASD/AFMDC Pd-4 \$/124/63/000/004/013/064

AUTHOR:

Karelin, V. STATE OF THE PERSON NAMED IN

TITLE:

Effect of hydrodynamic conditions and temperature of liquid upon the

initiation and development of cavitation

PERIODICAL: Referationy, zhurnal, Mekhanika, no. 4, 1963, 54, abstract 4B356

(Sb. tr. Mosk, inzh.-stroit. in-t., no. 40, 1962, 49-55

TEXT: The author has investigated the effect of the hydrodynamic characteristics and temperature of a liquid upon the amount of pressure required for the onset of cavitation; he also measured the pressure within a cavity at varying stages of cavitation. Tests were conducted in a Venturi tube, pressure was measured with piezometers placed on the tube's walls. The dependence (of the critical pressure needed for the onset of cavitation) upon the hydrodynamic conditions and the liquid's temperature is given in the form of curves. For cold water the amount of critical pressure is higher than that of the liquid vapors at a given temperature and approaches the pressure of the vapors as the temperature rises. B. S. Kogarko.

[Abstracter's note: Complete translation.]

Card 1/1

46683-66 EWT(1)/EWP(m)

ACC NR: AP6020733

SOURCE CODE: UR/0421/66/000/003/0120/0128

AUTHOR: Vulis, L. A. (Leningrad, Alma-Ata); Karelin, V. Ye. (Leningrad, Alma-Ata); ORG: none

TITLE: Propagation of a turbulent gas jet in a co-moving stream

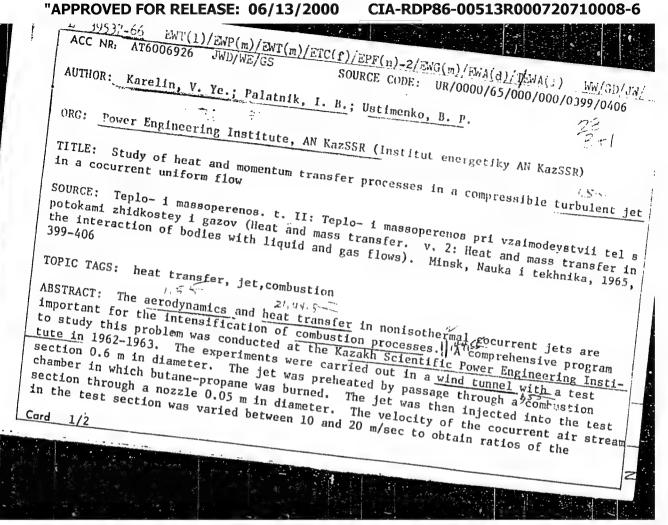
543

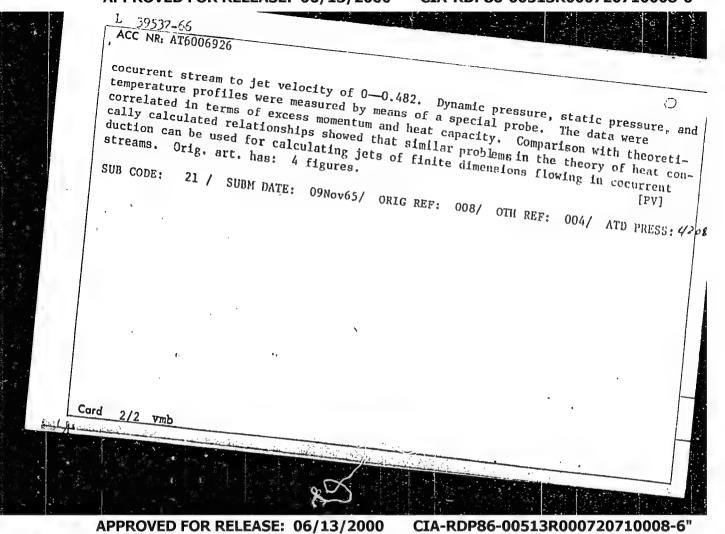
SOURCE: AN SSSR. Izvestiya. Mekhanika zhidkosti i gaza, no. 3, 1966, 120-128

TOPIC TAGS: axisymmetric flow, gas jet, turbulent jet, flow profile

ABSTRACT: The authors report the results of a detailed experimental investigation carried out in 1962-1964 on the laws governing the propagation of an axisymmetic jet of gas, heated slightly above the temperature of a stationary homogeneous medium, at small Mach numbers M < 1, at dynamic head ratios 0 < m < 0.23, velocity ratios of comparts of a stationary homogeneous medium, at 1.2 < m < 4.3. The experiments were made at different characteristics of compressibility (gas density ratio in the jet and in the surrounding medium) and co-motion (ratio of dynamic heads in jet and surrounding medium). The tests consisted of measuring the dynamic pressure head and the in an open wind tunnel of 0.6 m dia. The jet nozzle had a 50 mm dia. The experimental results are compared with calculations based on the method of the equivalent contradictory opinions regarding the effect of compressibility on the structure of the gas jet, special experiments were set up in which the initial turbulence level

Card 1/2





VULIS, L. A. (Leningrad); KARELIN, V. Ye.; PALATNIK, I. B.; SAKIPOV, Z.; "Laws of propagation of turbulent compressible gas jets" report presented at the 2nd All-Union Congress on Theoretical and Applied

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

L 19357-66 EWP(m)/EWT(1)/EWA(d)/EWA(1) ACCESSION NR: AT5011660 AUTHOR: Karelin, V. Ye. UR/3149/64/000/001/0006/0017 Service Services TITLE: Application of the method of the equivalent problem in heat conduction theory to the calculation of a nonisothermal axially symmetric turbulent jet within an 9 SOURCE: Alma-Ata. Kazakhskiy nauchno-issledovatel'skiy institut energetiki. Problemy teploenergetiki i prikladnoy teplofiziki, no. 1, 1964, Prikladnaya teplofizika, 6-17 TOPIC TAGS: axially symmetric current, aerodynamics, nonisothermal turbulent jet, current parameter comparison, equivalent heat conduction problem, thermal conductivity, associated stream ABSTRACT: Increased interest is being displayed in the laws of motion and turbulent transfer for nonisothermal jets generated within an associated current. However, there are no reliable measurements covering a sufficiently wide range of parameters $m_u = u_{current}/u_{et}$ and $\omega = y_{current}/y_{et}$ are no remains measurements covering a summerously wide range of parameters are no remain measurements covering a summerously wide range of parameters Yu. V. Ivanov, Izvestiya AN EstSSR, seriya fiziko-matematicheskikh nauk, 1962, no. 3;

L 19357-66

ACCESSION NR: AT5011660

O. Pabst, Luftfahrttechnik, 6 (1960), no. 10). Consequently, appropriate experiments were carried out at the Kazakhskiy nauchno-issledovatel skiy institut energetiki (Kazakh Scientific Research Institute for Power Engineering) during 1962-1963. The present paper reports some results of this study and compares it with the values calculated using the method of an equivalent heat conductivity problem developed by L. A. Vulis (Izvestiya AN KazSSR, seriya energeticheskaya, 1960, no. 2(18); L.A. Vulis, I. L. Senderikhina, Izvestiya AN KazSSR, seriya energeticheskaya, 1962, no. 1(22)). The calculation of the current's field is followed by a description of the experimental device (thermal current generated by the combustion of a butane-propane gas mixture), and numerous graphs comparing the experimental and theoretical values of various jet and current parameters (flow velocity, temperature, density, their ratios, excess density of momentum current, excess density of heat content current, etc. for isothermal and nonisothermal flow). "The author thanks L. A. Vulis and B. P. Ustimenko for directing the investigation." Orig. art. has: 7 formulas, 5 figures, and 1 table. ASSOCIATION: nome

Card 2/3

Card 3/3 60

KARELIN, V. Ye.; PALATNIK, I. B.; USTEMENKO, B. P.

"An investigation of heat and momentum transfer processes in a compressible turbulent jet in a uniform flow."

report submitted for 2nd All-Union Conf on Heat & Mass Transfer , Minsk, 4-12 Power Inst, AS KazSSR.

WARRIAN V.Ya. We of the method of the equivalent problem in heat includestion theory in maintaining a new containing antiquine tribal hardward jet in a state of the equivalent problem. I print here includes in a feet of the equivalent problem. I print here includes in a feet of the equivalent problem. I print here includes in a feet of the equivalent problem. I print here includes the equivalent problem in heat includes the equivalent problem. I print here includes the equival

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

KARELIN, Y. A.

"Mater Supply and Sewerage in Petroleum Refineries," by V. V. Abramov and Y. A. Karelin, Moscow-Leningrad, Gostoptekhizdat (State and Technical Publishing house of Petroleum and Mineral Fuel Literature), 1948. (Vodosnabzheniye i

KARELIN, Y A. A.

Technology

(Mater supply and canalization in the oil fields). Moskva, Gostoptekhiedat, 1951.

Monthly List of Russian Accessions, Library of Congress, Movember 1952. UNCLASSIFIED.

KARELIN, Ya.A.; ABRAMOV, V.V., inzhener, retsenzent; TOLOCHEO, M.M., inzhener, retsenzent; KOHTUSHKOV. A.M., redaktor

[Purifying industrial Sewage of the petroleum industry] Ochistka proizvodatvennykh stochnykh vod predpriiatii neftianci promyshlennosti. Moskva, Gos. nauchno-tekhn, izd-vo neftianci i gornotoplivnoi lit-ry, 1953. 295 p.

(Petroleum industry) (Waste products)

KARELIN, YA. A.

YAKOVLEV, S.V., kandidat tekhnicheskikh nauk; KARELIN, Ya.1.; MASLENNIKOV, N.A.; SHTEKKER, G.A., inzhener, redaktor; GOLDBERGUVK, L.A., redaktor; DAKHNOV, V.S., tekhnicheskiy redaktor

[Auxiliary installations in sewage purification stations] Vspomogatel'nye ustroistva ochistnykh kanalizatsionnykh stantsii. Pod.
red. S.V.IAkovleva. Moskva, Gos. izd-vo lit-ry po stroitel'stvu i
arkhitekture, 1955. 176 p.
(Sewage--Purification)

KONYUSHKOV, Andrey Maksimovich; YAKOVLEV, Sergey Vasil'yevich; ABRAHOV,
N.N. doktor tekhnicheskikh nauk, professor, retsenzent; KARELIH,
YA.A., kandidat tekhnicheskikh nauk, detsent, retsenzent; ZAMEVSKIY,
M.S., dotsent, redaktor; SMIRNOVA, A.P., redaktor; MEDVEDEV, L.Ya.,
tekhnicheskiy redaktor.

[Water supply and sever systems]Vodosnabzhenie i kanalizatsiia.
Moskva, Gos.ixd-vo lit-ry po stroitel'stvu i arkhitekture, 1955.
526 p.

(Water-supply engineering) (Sewerage)

(Water-supply engineering)

KARELIN, YA, A.

USSR, Chemical Technology. Chemical Products and Their I-12
Application--Water treatment. Sewage water

Abs Jour: Ref Zhur-Khimiya, No 3, 1957, 9167

Author : Karelin, Ya. A.

Inst : Nor given

Title : The Purification of Waste Waters from Refineries

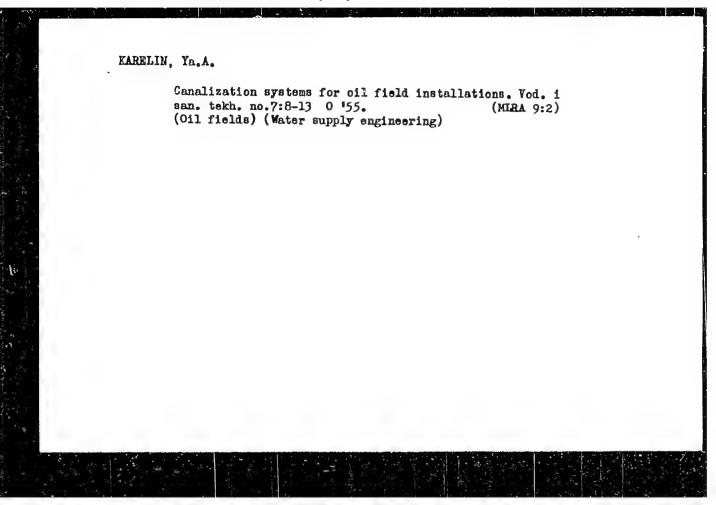
Orig Pub: Vodosnabzheniye i san. tekhnika, 1955, No 2, 23-26

Abstract: The water consumption in refineries per ton of

petroleum processed attains 10-80 m³; if the requirements of the plant steam heat electric power station are taken into account, the figure attains 120 m³. Of that total the condensation and cooling of the petroleum products account for 84%; the barometric condensers, 0%; washing requirements, 5%, other needs, 3%. The recycling of the standard-pure water and of part of the sewage water (SW) is an absolute necessity. The use of completely

separate piping systems with separate grids for the

Card 1/2



KARE LIN YA

Subject

: USSR/Engineering

Card 1/1

Pub. 78 - 18/27

Author

: Karelin, Ya. A.

Title

Design of shore installations for the intake from tankers and oil-carrying barges of petroleum wastes.

Periodical

: Neft. khoz., v. 33, #12, 71-77, D 1955

Abstract

In order to prevent the contamination of water in harbors by petroleum wastes, special installations must be built for the intake and disposal of the drained residues from oil-carrying tankers and barges. The layout of such installations is described. Diagrams, 3 references, 2 Russian, 1954 and 1955.

AID P - 3973

Institution: None

Submitted

No date

KARELIN, Ya.A.

Mechanical and chemical purification of industrial waste water of a Philadelphia refinery (From "The Petroleum Engineer" ne.11, 1954).

Ved.i san.tekh.ne.5:32-35 My '56. (MLRA 9:9)

(Philadelphia--Water--Purification)

· restlie 1/4 1

USSR Chemical Technology. Chemical Products

H-5

and Their Application

Water treatment. Sewage water.

Abs Jour: Referat Zhur - Khimiya, No 1, 1958, 1792

Author : Karelin Ya. A., Belinskiy M.L.

Title : Sewer Systems at Petroleum Production Bases

Orig Pub: Vodosnabzheniye i san. tekhnika, 1956, No 11.

13-17

Abstract: At petroleum production bases 2 sewer systems

are planned: an industrial and storm sewer system and a household system. In the industrial and storm sewer system are installed 2 sectional petroleum traps, from which the sewage water passes into ponds. If the sewage water contains tetraethyl lead the latter is extracted with the lightest aviation gasoline containing no ethyl

Card 1/2

USSR /Chemical Technology. Chemical Products and Their Application

H-5

Water treatment. Sewage water.

Abs Jour: Referat Zhur - Khimiya, No 1, 1958, 1792

lead. After extraction the sewage water is allowed to settle for 10-20 how &,

Card 2/2

KARRLIN, Ya.A

Layouts of sewer systems for washing and steaming centers of railroad stations. Vod.1 san.tekh. no.6:9-13 Je '57. (MLRA 10:7) (Sewage disposal)

KARCLIN, YA A.

AUTHORS: Karelin, Ya.A. and Vorob'yeva, G.I. 65-10-6/13

Biochemical Purification of Effluent Waters from Refineries TITLE:

(Biokhimicheskaya ochistka stochnykh vod nefteperer-

abatyvayushchikh zavodov)

PERIODICAL: Khimiya i Tekhnologiya Topliva i Masel, 1957, No.10, pp. 29-34 (ÚSSR)

ABSTRACT: Laboratory investigations on the possibility of the application of bacteriological purification of refinery effluents are described. It was established that Pseudomonas bacteria actively decompose crude oil and individual hydro-Or the cultures separated, the most active were: carbons. Pseudomonas Putida Flugge, 1886; Pseudomonas Dacunhae (Gray and Thornton, 1928) and an undetermined type which was called Pseudomonas species. There are 2 figures, 6 tables and 5 references, 1 of which is Russian and 4 English.

ASSOCIATION: MISI imeni V.V. Kuybysheva

AVAILABLE:

Library of Congress

Card 1/1

KARElin, Yt. A.

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6"

Karelin, Ya. A., a specialist with the MISI imeni AUTHOR:

Kuybyshev

TITLE:

How to Improve Refinery Water Supply and Sewer Systems (Usovershenstvovaniye sistem vodosnabzheniya 1 kanalizatsii na neftepererabatyvayushchikh zavodakh)

PERIODICAL:

Neftyanoye Khozyaystvo, 1957, Nr 5, pp. 53-58 (USSR)

ABSTRACT:

The successful operation of a modern Soviet refinery requires hundreds of millions of cubic meters of water. Refineries producing mainly fuels require from 30 to 40 cubic meters of water per ton of crude oil, when the temperature difference between the incoming and outgoing waters is 25°C. Refineries geared for the production of fuels and lubricants require 50 to 60 cu. m of water per ton of crude oil put through and refineries producing a greater variety of products require 70 - 80 cu. m. If the heat and power plant is included, the water requirements are approximately 100 cu. m. per ton of crude oil. On the average 92% of water is used for condenser-cooling

Card 1/6

How to Improve Refinery Water Supply and Sewer Systems 93-5-14/19 (Cont.)

purposes. Since in this case, as a rule, there is no direct contact with any product, the water is considered clean and can be recirculated in the system. Only 3% of the water is used for condensation through direct contact. In refineries processing sour crude the water used for condensation through direct contact becomes contaminated with hydrogen sulfide. Of the remaining 5% of water, 3% is used to replace evaporation losses and 2% for washing purposes. 96.4% of the water required by heat and power stations is used for indirect contact cooling and the remaining 3.6% for other purposes. The Glavneftepererabotka refineries report that only 48% instead of 94-96% of water is put again into the system, while 12 refineries use only fresh water. Every effort should be made to reduce the quantity of water discharged into the sewers and natural water reservoirs. New refineries provide for two types of water recirculation systems. One type is for equipment used in the refining of crude and heavy petroleum products and the other for equipment used in processing gas (C5 and lighter) and light petroleum products as well as for lubricating oil cooling units and compressor stations. Provisions are also made for a recir-Card 2/6

How to Improve Refinery Water Supply and Sewer Systems 93-5-14/19 (Cont.)

culating water systems for condensers of the atmospheric - vacuum pipestills. A repeated use of water is recommended for new refineries and for those which are being reconstructed. V. Ya. Myagkov and Ya. G. Sorkin, co-authors of an article dealing with methods of improving the use of water and heat at refineries, are referred to as men who realize the importance of properly utilizing water and heat for refinery purposes. The selection of a sewage system should be dictated by economic considerations and local conditions. In this connection the problem of purifying industrial waste waters is of prime importance. systems are proposed for the efficient operation of refineries: 1) A sewage system whose waters can be reused after treatment and 2) a sewage system whose treated waters are discharged. The following waters go into the first sewage system: a) water used for washing the refinery equipment and tanks, b) waste water from condensers and scrubbers except water from atmospheric (vacuum pipestill condenser) c) storm waters from various platforms and storage tank farms except crude oil storage tank farms. Water treating facilities of the first sewage system consist Card 3/6

How to Improve Refinery Water Supply and Sewer Systems (Cont.)

of grills, sandtraps, oil traps in which the water is to stand for two hours, ponds in which additional settling is to take place (6-24 hours) and a reserve tank with a capacity equal to a 3-day volume of waste waters, sand filters designed by I. L. Mongayt and I. D. Rodziller, and a collector tank with a capacity equal to a 2-hour consumption of water. The water is recirculated in the first system. The second sewage system consists of the following sewage networks: (1) a sewer network collecting waste waters containing emulsified oils from electrical desalting units, crude oil tank farms and deasphalting unit condensers; 2) a network collecting waste waters containing alkali sulfides; 3) a network collecting waste waters containing acids and sulfates; 4) a separate network collecting waters which require special treatment. Waste waters of this system undergo preliminary treatment and purification as they pass through various oil traps, sand traps, grills and other devices similar to those used in the first sewage system. Certain waste waters require, however, additional treatment like flotation, deodorization, neutralization, biochemical treatment and are then channeled to a large reservoir wherefrom several days later they are Card 4/6

How to Improve Refinery Water Supply and Sewer Systems 93-5-14/19 (Cont.)

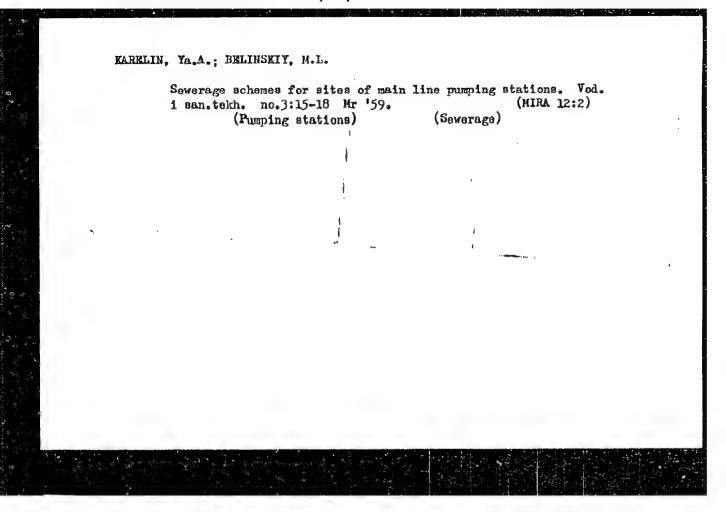
action of the reagents water is separated from oil. The top layer, a relatively pure oil, is pumped to the crude oil storage tanks. The medium layer consisting of water, soluble matters and oil, goes again through the second dehydration stage. Organic reagents are used this time and the mixture is heated again, and allowed to separate. The recovered oil is pumped to crude oil storage tanks. The bottom layer representing heavy petroleum products, water and solid matters should be drained from the tanks and burned in special furnaces. Large refineries recovering considerable quantities of oil from oil traps should have an independent treating plant. A basic scheme of water supply and sewerage system has been worked out by the following specialists: V. V. Abramov (Giprospetsneft'), S. I. Beletskiy (Giproneftezavod), N. M. Litvishkov (Giproazneft'), P.A. Mikheyev (Giprogrozneft'), S. D. Klimov (Giprogaztopprom), B. A. Mitkalev (UfNII) and Ya. A. Karelin (MISI imeni V. V. Kuybyshev). The advantage of this scheme is as follows: The quantity of industrial waste waters discharged into natural bodies of water is reduced to a minimum by the use of the above mentioned methods of treating the refinery waste waters. This reduction of waste waters cuts down on the pollution of natural waters. There are 2 figures.

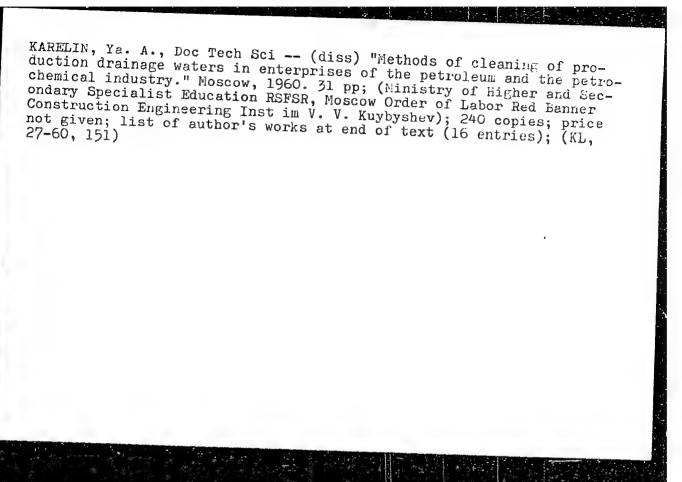
ASSOCIATION: MISI imemi Kyubysheva AVAILABLE: Library of Congress

Card 6/6

KARELIN, Yakov Aleksandrovich; KONYUSHKOV, A.M., red.; L'VOVA, L.A., vedushchiy red.; POLOSINA, A.S., tekhn.red.

[Purification of waste waters from oil fields and petroleum refineries] Ochistka stochnykh vod neftianykh promyslov i zavodov. Moskva, Gos.nauchno-tekhn.izd-vo neft. i gorno-toplivnoi lit-ry, 1959. 343 p. (MIRA 11:11) (Petroleum waste)





KONYUSHKOV, Andrey Maksimovich, kand.tekhn.nsuk; YAKOVLEV, Sergey
Vasil'yevich, doktor tekhn.nsuk. Prininsl uchsstiye FEDOROVSKIY,
N.A., inzh. AERAMOV, N.N., prof., doktor tekhn.nsuk, retsensent;
KARELIN, Ya.A., dotsent, kand.tekhn.nsuk, retsensent; ZANEVSKIY,
M.S., dotsent, nsuchnyy red.; SMINMOVA, A.P., red.izd-va;
EL'KINA, E.M., tekhn.red.

[Vater-supply and sewerage] Vodosnabzhenie i kanslizatsiis. Izd.2.,
ispr. Moskva, Gos.izd-vo lit-ry po stroit., arkhit. i stroit. materialam, 1960. 534 p.

(Water-supply engineering) (Sewerage)

(Water-supply engineering)

SHISHKIN, Zakhar Nesterovich; KARELIN, Yakov Aleksandrovich, dotsent; KOLOBANOV, Sergey Konstantinovich, dotsent, kand.tekhn.neuk; YAKOVLEV, Sergey Vasil yevich, doktor tekhn.nauk; ZHUKOV, A.I., prof.; GULYAYEV, N.F., kand.tekhn.nauk; SUKHIY, P.A., inzh., retsenzent; POPOVA, N.M., kand.tekhn.nauk, retsenzent; SMIRNOVA, A.P., red.izd-va; GILENSON, P.G., tekhn.red.; TEMKINA, Ye.L., tekhn.red.

[Sewerage] Kanalizatsiia. Izd.2., ispr. Pod red. A.I.Zhukova. Moskva. Gos.izd-vo lit-ry po stroit., arkhit. i stroit.materialam. 1960. 592 p. (MIRA 14:4)

KARELIN, Yekov Aleksandrovich; PERSYALOV, Vyacheslav Georgiyevich;

SMIRKOVA, A.P., red. izd-ve; OSEIKO, L.M., tekhn. red.

[Removal of petroleum products from waste waters; foreign practices] Ochistke stochnykh vod ot nefteproduktov; zerubezhnyi oppt. Moskva, Gos. izd-vo lit-ry po stroit., arkhit. i stroit, meterialem, 1961. 130 p. (MIRA 14:5)

(Sewage--Purification)

(Unite States--Petroleum industry--Water supply)

MARELIN, Ya.A.; kand.tekhn.nauk

Disposal of petroleum refinery waste waters. Zhur. VKHO 6 no.2:166172 '61.

(Sewage disposal) (Petroleum refineries)

(Sewage disposal) (Petroleum refineries)

Experimental investigation of the two-stage blochemical purification of the waste waters of electric desalters of the Orek Petroleum Refinery. Khim. i tekh. topl. i masel 6 no.11:23-27 N '61.

1. Moskovskiy inzhenerno-stroitel'nyy institut im. V.VIKuybhsheva i Orskiy neftepaperabatyvayushohiy zavod.

(Qrsk-Petroleum waste-Purification)

S/065/61/000/008/006/009 E030/E535

AUTHOR: Karelin, Ya.A.

TITLE: Improvement of refinery waste disposal systems by

surface-active agents

PERIODICAL: Khimiya i tekhnologiya topliv i masel, 1961, No.8, pp.36-41

TEXT: The waste water from the electrolytic desalting and ATK washing plants had an unsatisfactorily high content of stable emulsions of petroleum products. This was attributed to the use of ionic surface-active sulphonates and HYK (NChK) that was supplied by the Yaroslav NPZ. In an effort to improve the condition of the waste water, the use of 0π -10 (OP-10) in the electrolytic refining units was tried during investigations carried out in November 1958 at the Novo-Gor'kiy neftepererabatyvayushchiy zavod (Novo-Gor'kiy Refinery). OP-10 is a non-ionic surface-active agent with the structure

Card 1/3

Improvement of refinery waste ...

\$/065/61/000/008/006/009 E030/E535

where n=10 to 12 and the R are hydrophobic alkyl groups with 9 to 10 atoms of C counter-balancing the hydrophilic polyethylene glycol groups. It is a light-yellow to brownish yellow paste with a mouldy smell with specific weight of 1.06 to 1.08 and pH of the aqueous solution (concentration 10 g/1) 6.8.

Improvement of refinery waste ... \$/065/61/000/008/006/009 E030/E535

The best use of OP-10 was found to be by addition as feed to the pumps in the waste system in the form of a 2.5% aqueous solution using 50 g/ton of petroleum product. The electric fields in the electrolytic units were 1175 V/cm in the first stage and 2350 V/cm in the second; the temperature was 80°C. Comparing corresponding monthly figures before and after the use of OP-10; a drop from 243 mg/l to 57 mg/l of petroleum products in the waste was observed. As a result, the use of OP-10 is proposed as cheaper and easier than that of ionic surface-active sulphonates; biochemical agents have still to be used for the purification process of the water (to eliminate the petroleum and mouldy smell of the surface-active agents used). There are 4 figures;

Card 3/3

KARELIN, Ya.A.

Biochemical purification of waste waters from the Electrical Desalting Unit using the OP nonionic surface-active agents. Khim. i tekh. topl.i masel 7 no.1:9-14 Ja 162. (MIRA 15:1)

1. Moskovskiy inzhemerno-stroitel'nyy institut im. V.V.Kuybysheva.
(Petroleum-Refining-Desalting)
(Sewage-Purification)

KARELIN, Ya.A.; BELINSKIY, M.L.

Sewer systems for sections of filling stations of petroleum products pipelines. Neft. khoz. 40 no.1:58-64, Ja '62. (MIRA 15:2) (Petroleum waste)

WARELIN, Ya.A., kand.tekhn.nauk; SUKHODOL'SKIY, A.M., inzh.

Use of combined structures for the purification of sewage by trickling. Vod. i san. tekh. no.6:38-40 Je '62. (MIRA 15:7) (Sewage—Purification)

ZHUKOV, D.D.; KARELIN, Ya.A.; MEDEM, V.M.; NAZAROV, I.I.; SHEVTSOV, D.A.

Additional experimental investigations of a two-stage biochemical purification of waste waters from the Electrical Desalting Unit of the Orsk Petroleum Refinery. Khim.i tekh.topl.i masel 7 no.9:19-23 S '62. (MIRA 15:8)

1. Moskovskiy inzhenermo-stroitel:nyy institut im. V.V.Kuybysheva i Orskiy neftepererabatyvayushchiy zavod. (Orsk--Petroleum--Refining) (Sewage--Purification)

KARELIN, Ya.A.; SOY(LOV, A.G.

Quality of the waters injected into producing reservoirs.

Neft.khoz. 41 no. 12:40-45 D 163. (MIRA 17:6)

ALEKSEYEVA, V.A.; KARELIN, Ya.A.

Removing dissolved petroleum from waste water using ozone. Nefteprom. delo no.4:33-35 63. (MIRA 17:8)

1. Moskovskiy inzhenerno-stroitel nyy institut im. Kuybysheva.

ALEKSEYEVA, V.A.; KARELIN, Ya.A.

Final purification of waste waters with ozone. Nefteper. i neftekhim. no.5:19-21 '63. (MIRA 17:8)

l. Moskovskiy inzhenerno-stroitel*nyy institut im. V.V. Kuybysheva.

1-16928-65 Pb-4/Pa-4 AFVIL/AND

ACCESSION NR: AP5002813

8/0065/64/000/008/0029/0037

AUTHOR: Karelin, Ya. A.; Ikramov, M.; Zhukov, D. D.; Komarov, D. Ye.

TITLE: Investigation of industrial waste waters of an oil refinery and their purification by a biochemical method

SOURCE: Khimiya i tekhnologiya topliv i masel, no. 8, 1964, 29-37

TOPIC TAUS: petroleum industry, waste disposal, water sanitation, biochemistry

ARSTRACT: Concise information is presented on the amount, composition and basic properties of the sewer waters of the Yaroslavi Oil Refinery imeni D. I. Hendeleyev and their biochemical purification. The sewer waters of this refinery were found to be characterized by the presence of petroleum and petroleum products, volatile and nonvolatile phencis, notrobenzene, fatty acids, paraffin, and sulfur compounds. In spite of the complex and varied composition of the organic contaminants contained in the sewer waters of the bil refinery, their biochemical purification was found to be quite possible. Data were obtained that can be used in the planning of industrial purification installations. The average oxidative capacity of two-stage seration installations in work on total purification should be taken to be 500 g/m at a concen-

Card 1/2

·L 16928-65

ACCESSION NR: AP5002813

tration of active sludge of & g/liter with respect to dry matter. Rifective operation of the aeration installations required the delivery of 50 cubic meters of air per kilogram of reduced biochemical oxygen consumption, at a depth of the aeration tanks of 4 meters. The duration of the seration period was found to depend on the degree of contamination and temperature of the water to be purified. For sewer waters analogous to those investigated, an aeration period of 18-24 hours, including 6-8 hours for the first stage and 12-16 hours for the second, is recommended. The temperature of the sewer water should be in the range of 20-28°C. Orig. Art. has 1 figure, 1 graph, and

ASSOCIATION: MIST

SUBMITTED: 00

NCL: 00

SUB CODE: GO, LS

NO REF SOV: 005

OTHER: 002

JPRS

Card 2/2

ZHUKOV, Aleksandr Ivanovich, prof., doktor tekhm. nauk; KARELIN, Yakov Aleksandrovich, prof.; KOLOBANOV, Sergey Konstantinovich, dots., kand. tekhm. nauk; YAKOVLEV, Sergey Vasil'yevich, prof.; LUKINYKH, N.A., kand. tekhm. nauk, retsenzent; MONGAYT, 1.L., kand. tekhm. nauk, retsenzent; SHKUNDIN, R.F., inzh., retsenzent; SKVORTSOVA, I.P., red.

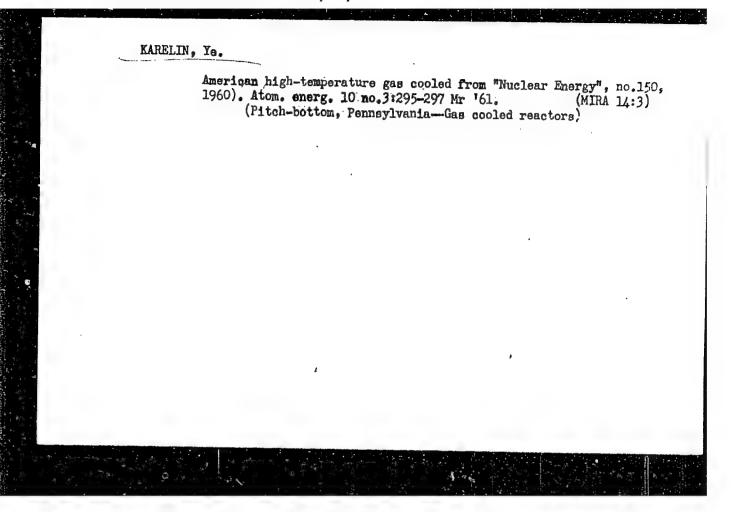
[Sewerage] Kanalizatsiia. Izd.3., ispr. i dop. Moskva, Stroiizdat, 1964. 641 p. (MIRA 18:2)

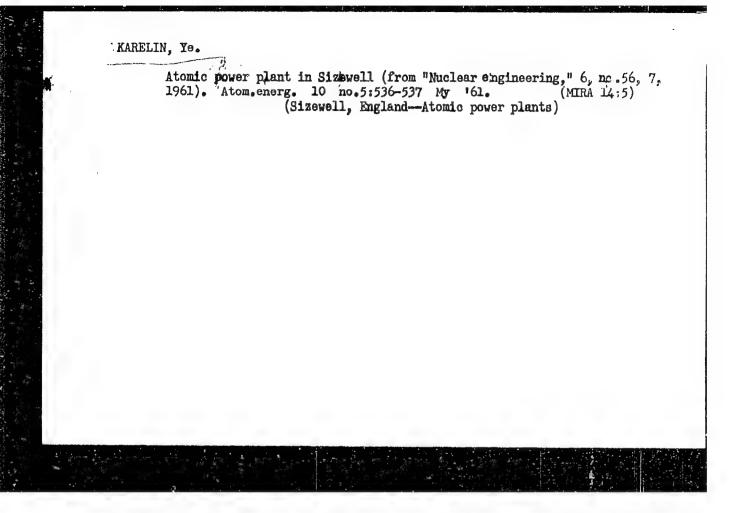
EARRELIE, Ya.A., doktor tekhn. nauk; EYAZANCY, V.L., inzh.

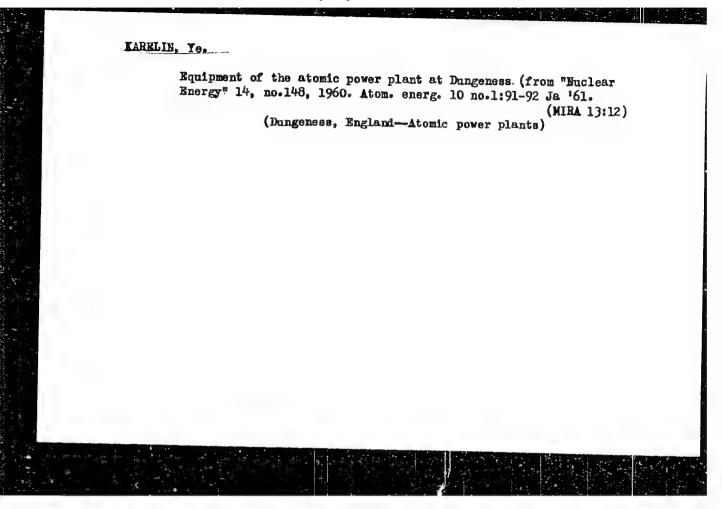
Horizontal settling tank with vertical water flow. Ved. i san.
tekh. no.12:35-36 D 163 (NIRA 18:2)

CIA-RDP86-00513R000720710008-6"

APPROVED FOR RELEASE: 06/13/2000







KARBLIN, Yu.

Nomogram for rapid determination of series capacitance connections or parallel resistance connections. Radio no.11: Supp.31

(Radio circuits)

(MIRA 10:10)

GREBENNIKOV, O.F.; MYASNIKOV, S.I.; KARELIN, Yu.A.; ZUBKOV, G.A.

Attachment to the 16S-2 "Kiev" motion-picture camera for semiautomatic control of the lens diaphragm. Trudy LIKI no.11:35-38 164.

(MIRA 18:10)

1. Kafedra kinofotoapparatury Leningradskogo instituta kinoinzhenorov.

316**2**0 \$/138/61/000/012/002/008 \$13.92.05 A051/A126

AUTHORS: Kartsev, V.N.; Karelina, G.G.; Rozova, N.I.

TITLE: Properties of siloxane rubber vulcanizates with a low content of

vinyl groups

2

PERIODICAL: Kauchuk i rezina, no. 12, 1961, 7 - 11

TEXT: Experimental results are submitted from an investigation of test samples of vinylsiloxane polymers with a low content of vinyl groups [CKTB (SKTV)], as compared to dimethylsiloxane rubber [CKT(SKT)]. The SKTV samples were produced on an experimental BHUMCK(VNIISK) equipment, using "acedic" (samples no. 1, 2, 226) and "alkaline" (sample no. 19) catalysts. The SKTV and SKT based mixes were produced on laboratory rollers, according to the following composition in weight parts to 100 weight parts of raw rubber:

	SKTV	SKT
silica gel y -333 (U-333)	50	50
zinc oxide		5
benzoyl peroxide paste (95% benzoyl peroxide and		
siloxane oil, in the ratio of 1:1)	1.26	4.2

Card 1/3

31620 \$/138/61/000/012/002/008 A051/A126

X

Properties of siloxane rubber vulcanizates with ...

It was found that vulcanizates based on a SKTV rubber mix containing silica gel U-333 and a lowered quantity of benzoyl peroxide (0.6 w.p.) are characterized by a reduced residual deformation and a higher thermal stability than vulcanizates of the standard SKT rubber mixes. The thermal stability of the SKTV-based mixes may be increased by replacing the zinc oxide with iron oxide or titanium dioxide. The SKTV and SKT vulcanizates do not differ in their tendency to destruction when heated in a closed system, at 200°C. They also have similar dielectric properties. The vulcanizates of the SKTV siloxane rubber, produced in the presence of the "acedic" and "alkaline" catalysts, were found to be the same in their main physico-mechanical characteristics. The SKTV vulcanizates, produced with disumyl peroxide or ditertiary butyl peroxide, as compared to vulcanizates containing benzoyl peroxide, were found to have a lower residual deformation and a much lesser tendency to destruction when heated without air. It was further found that mixes containing channel black, do not vulcanize, even in the presence of increased amounts of dicumyl peroxide or ditertiary butyl. In the case of furnace carbon black, vulcanizates were obtained with satisfactory properties. The SKTV vulcanizates containing the furnace carbon black and the ditertiary butyl peroxide are equivalent to vulcanizates based on the same rubber, containing the U-333 silica gel, but the former do have in-

Card 2/3

31620 \$/138/61/000/012/002/008 A051/A126

Properties of siloxane rubber vulcanizates with ...

creased residual deformation after compression. Tests for electroconductivity of the SKTV vulcanizates containing the furnace carbon black indicated that these rubbers are semi-conductors (specific volumetric electrical resistance is equal to 1.0 x 10⁶ ohm/cm). There are 8 tables, 2 figures and 4 references: 1 Soviet-bloc and 3 non-Soviet-bloc. The reference to the most recent Engish-language publication reads as follows: G.M. Konkle, R.M. Savage, Rubb. Age, no. 6, 975 (1959).

ASSOCIATION: Nauchno-issledovatel'skiy institut sinteticheskogo kauchuka im. S.V. Lebedeva (Scientific Research Institute of Synthetic Rubber im. S.V. Lebedev)

λ.

Card 3/3

BORISOV, S.N.; KARELINA, G.G.

Dependence of the properties of rubber made from vinyl siloxanes on the vinyl group content of the elastomers. Kauch. i rez. 22 no.6:6-10 Je '63. (MIRA 16:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo kauchuka im. S.V. Lebedeva. (Rubber, Synthetic-Testing) (Siloxanes)

KARELINA, GG.

3

18.8310

26908

8/138/61/000/009/co1/005 4001/4129

AUTHORS:

Klebanskiy, A. L., Tsukorman, N. Ya., Kartsev, V. N., Labatin, A. L., Trenko, Yu. V., Malishina, L. P., Borovikova, N. A., Karelina, G. C., Rozhkov, Yu. P.

TITLE:

A new type of chloroprene rubber: liquid nairite (This work was awarded the second prize at the VahO im. D. I. Mendele-yev compositions in 1959)

PERIODICAL: Kauchuk i rozina, no. 5, 1961, 1 - 5

The right chemical stability, the gasoline-potrolsum stability and ozone-resistence of chloroprene rubber makes it a suitable material for anti-corresion coating and hermotic scaling. However, the difficulty of producing highly-concentrated solutions based on commercial nairite limited the application of the latter in anti-corresion technique. It has been assumed that the use of low-molecular polymers for this purpose would enable one to obtain low-viscous, highly-concentrated solutions satisfying the anti-corresion techniques. One of the matheds for producing low-molecular polymers is the use of the polymerization of increased concentrations of regulator-compounds able to break the chains and to form new ac-

Card-1/6-

3

A new type of chloroprene rubber: liquid nairite

S/138/61/000/005/001/006 A051/A129

tive centers. Sulfurous compounds, such as mercaptane, threacids, mentingcasculfides, are widely used as regulators. When studying the action of n-tetradecylsercaptano, diisopropylxanthegomedisulfide and bisethylmanthegomediculfide during the process of polymerization of chloroprene, it was established that with an increase in the concentration of the regulator the molecular weight of the polymor drops correspondingly and the plasticity of the rubber increases. It was also ad that the use of greater quantities of bisethylmenthogenediculfide in the polymerimetion of chloroprene in emilsion decreases the molecular weight of the polymer and yields low-viscosity solutions of rubber. An attempt was made to produce low-molecular polychloroprone by polymerization of chloroprene in the presence of sulfur with subsequent destruction of the polymer. It was shown that the action of sulfur differs from that of other regulators. The effect of sulfur on the polymers of chloroprone is shown by the scheme: $-(\text{CH}_2-\text{CCl}+\text{CH}-\text{CH}_2)_n-\text{S}_x-(\text{CH}_2-\text{CCl}+\text{CH}-\text{CH}_2)_n-\text{S}_n$, where x=2-6. The sulfur forms linear bonds in the polymer chain. With in increase in the bound sulfur content in the polymer the molecular weight of the polymer decreases in the subsequent interaction with thiuram from 600,000 to 280,000 with 0.35 of bound sulfur and from 300,000 to 43,000 with 1% of bound sulfur. The quantity of reacted thiuram increases respectively. The destruction scheme is given as follows: 1) The formation of free radicals under the effect of the thermal action or thiurans

Card 2/6

3

26983

S/128/61/000/005/001/006 A051/A129

A new type of chloroprene rubbor: liquid nairite

 $-(cH_2-cc1+cH_2)_n-s-s-s-(cH_2-cc1+cH_2)_m-s-s-s-s- \to -(cH_2-cc1+cH_2)_n-s;$

2) Recombination of the polymer radical with molecular thiuram and splitting : off of the latter along the -S-S-bond:

$$\hbox{-(cH$_2$-cc1$-cc+cH$_2)$_n$-s} \cdot + \hbox{(c$_2$H$_5)$_2$N-c$-s-s-c$-N(c$_2H_5)$_2} \to \\ \hbox{$\overset{\circ}{\mathbb{S}}$} \quad \hbox{$\overset{\circ}{\mathbb{S}}$} \quad \hbox{$\overset{\circ}{\mathbb{S}}$}$$

$$-7 - (CH_2 - CC1 = CH - CH_2)_n - S - S - C - N(C_2H_5)_2 + (C_2H_5)_2 - C - 3 - (CH_2 - CC1 = CH - CH_2)_n - (CH_2 - CC1 = CH_2)_n - (CH_2 - CH_2)_$$

Based on the outlined assumptions of the mechanism of the bulfur action during the process of chloroprene polymerization and destruction of the polymer under the offect of the chemical masticating substances, the conditions for producing low-molecular chloroprene rubber-"liquid" natrite were developed. The liquid types of natrite can be obtained on a typical apparatus. The sulfur can be introduced in the form of solutions in mineral oils as well as aqueous dispersions obtained in the presence of emulsifiers and protective colloids. It was shown by V. N. Kartsev, M. A. Gutman, G. G. Karelina, F. Ye. Berman, Ye. G. Malinovskaya, M. B. Shur at WHISK, no. 2389, 1951, that for mastication the most effective system is morcapto-

Card 3/6

3/

A new type of chloroprene rubber: liquid mairite

8/139/61/000/005/001/005 A051/A109

beneathiczol (captam)-diphenylguanidine (DPhS). To increase the activity of these agents, tetramethylthiurameteulfide was edded (thiuram D) or tetracedelthiuramilsulfilde (thiuram E). Literature data indicate that active morticating against of polychloroprene are the piperidine salt of hexamothylonedithicearb; sine said or amagnium hemmethylenedithiosarbamate. The order of introduction of the agents plays an important role. The effect of the type and composition of the carbon black on the solubility of the rubber mixtures from "liquid" mainte and investigated. Only the thermal carten black helps to retain couplete solubility. Higher indices of relative elengation when filling with 100 m.p. and over and achieved with thermal carbon black. The composition and technology for proparing the rubber mixtures based on the "liquid" mairite with thermal carbon black as faller yielded highly-concentrated solutions (70 - 75%). These solutions are suitable for scaling various equipment by the same methods which are used in the case of dye and varnish coatings. Tests of coatings made of liquid mainite in emperimental and natural samples in various industrial fields showed the empediancy of using this product as a material for protecting the metal from corresion; cresion, cavitation and also as a material for hermatic scaling. There are 4 tables and 21 references: 2 Soviet-bloc, 19 non-Soviet-bloc. The references to the 4 most recent

Card 4/6

"APPROVED FOR RELEASE: 06/13/2000 CIA

CIA-RDP86-00513R000720710008-6

3

A new type of chloroprene rubber: liquid nairite

S/138/61/000/005/001/006 A051/A120

Publish-language publications read as follows: Corros. Technol., 5, no. 4, 107 (1958); R. B. Seymour a. oth., Plastics for Corrosion Resistant Application, M.Y., 1955, 90; Rubb. a. Plast. Age, 39, no. 8, 684 (1958); Corros. Technol., 3, no. 3, 89 (1956).

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel skiy institut sinteticheskogo kauchu-ka im. S. V. Lebedova (AE-Union Scientific Research Institute of Synthetic Rubber im. S. V. Lebedov)

Card 5/6

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000720710008-6

KARELINA, G.G.

27514 5/138/61/000/005/002/005 805144129

15-9201

AUTHORS:

Labutin, A. L., Klebanskiy, A. L., Tsukerman, N. Ya., Kartsov, V. N., Trenko, Yu. V., Kal'shina, L. P., Borovikova, N. A., Karelina, G. G., Roshkov, Yu. P.

tominov, iu.

TITLE: "Liquid nairito" - a new material for rubberizing

PERIODICAL: Kauchuk i rezina, no. 6, 1961, 5 - 8

The authors state that in the chemical destruction of "liquid" nairite, highly concentrated solutions can be produced which are applicable as a natorial for rubberizing. In the USSR a safer binary solvent, consisting of 2 weight parts of ethylacetate and 1 w.p. of gasoline is used in nairite adhesives. Experiments showed, however, that this solvent in "liquid" nairite is not suitable for many technical reasons. Better results were obtained in using a termary solvent consisting of 76% solvent, 19% turpentine and 5% n-butanel. The latter compensated does not dissolve the mairite, but facilitates the use of the brush for painting and good coating distribution. It was noted that film vulcanization from liquid nairite at 20°C does not show positive results. Thus various forms of thermal vulcanization were investigated; vulcanization with heated air, live vapor, het water

Card 1/6

4

275ldi \$/138/61/000/005/002/006 A051/A129

"Liquid mairite" - a new material for rubberizing

and infra-red irradiation. It was established that the most suitable method was vulcanization by hot air. The physico-mechanical indices of nairite centings vulcanized in air at various temperatures are given in Fig. 1. Fig. 2 shees the relationship between the temperature and duration of the vulcanization. The most suitable temperatures of vulcanization in air are within the range of 100 - 142°C. It was noted that the liquid mairite coatings did not pessess the proper adhesion to metal. Thus certain other adhesives or coatings ensuring better adhesion between metal and coating were sought. The best results were obtained with the following three materials; standard louconate (organic base; n, n, n, n - triisocyanatetriphenylmethane), chloronairite adhesive (organic base: chloronairite and nairite) and a primer, tentatively called opoxide primer (organic base: epoxide resin, chloronairite and nairite). The chemical stability and anti-corrosion properties of the vulcanized mairite coatings were studied. The conclusion was drawn that 1.2-mm nairite coatings in combination with a water-resistant coating applied three times can reliably protect motals from corrosion due to aqueous solutions of many acids, alkali and salts. The coatings were not resistant to the action of oxidizing agents, aromatic and halided solvents. Embber coatings differ from varnish and plastic coatings by an increased resistance to abrasive wear. An attempt was made

Card 2/6

275hh \$/138/61/000/006/002/006 A051/A129

"Liquid nairite" - a new material for rubberizing

to determine the resistance of mairite coatings under conditions of dry friction using the Orossolli-type machine. It is concluded that coatings of so-called crystallizing liquid mairite obtained in low-temperature polymerization are surerior to other rubbers in their wear-resistance, excepting vulcollane, which has a unique resistance to abrasive wear. It was established that coatings of liquid oil. nairite are superior to coatings of bakelite, polyothylene and caprone, when tested in rapidly flowing sea water. Tests have further shown that liquid mairite as a material for coatings will become widely used in industry in the next few years. At present tests are being conducted in the North Sea and the Atlantic Ocean on propellers of fishing trawlers coated with liquid nairite for protection from corrosion, erosion and cavitation. Mechanical plants are testing steel covers of refrigerators and condensators coated with nairite. These were previously manufactured from non-ferrous motals. Cortain chemical plants have installed diaphragm valves, the interior of which is covered with liquid nairite to prevent correction from acid solutions, alkali and salts. The possibility of using nairite coatings in various instruments as a means for preventing spark formation in percussion has also been revealed. Finally, it was established that these coatings can be used in certain constructions for hermetic scaling. At the Moscow TERS NO 12 a vacuum-condensator of a mass-produced 50 thousand kw steam turbine withstood a

Card 3/6

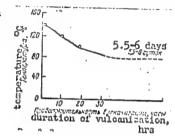
275\\\\ 8/138/61/600/005/602/005 A051/A129

"Liquid mairite" - a now material for rubberizing

testing period of one and a half years with the brass pipes and steel pipe boards coated with liquid nairite. K. S. Shmurey, O. P. Abolina, A. I. Konstantinova and G. A. Selivanovskaya took part in the work. There are 2 tables and 2 sets of graphs.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo kau-k-chuka im. S. V. Lebedeva (All-Union Scientific Research Institute of Synthetic Rubber im. S. V. Lebedev)

Fig. 2. Dependence of the vulcanization duration of the coatings made of liquid nairite on the temperature



Card 4/6

37175

S/138/62/000/004/001/008 A051/A126

15.9209

Korotkina, D.Sh.; Vinogradova, V.V.; Karelina, G.G.

TITLE:

AUTHORS:

Copolymerization of unsaturated phosphor-organic compounds

PERIODICAL:

Kauchuk i rezina, no. 4, 1962, 1 - 3

TEXT: The effect of the phosphorous atom on polymer properties was investigated and a comparison was made of the sodium-butadiene and acryl rubber properties with those of similar polymers containing phosphorous in the side chain. The ethers of allyl-, butadiene-, isoprene-styrene-phosphene acids were used as the phosphorous-containing monomers in the experiments. The Φ3K- M (FEK-M) photocolorimeter was used to determine the phosphorous content in the initial products and polymers. The introduction of the phosphorous atom into the polymer chain of the sodium-butadiene rubber was found, in most cases, to improve considerably the physico-mechanical properties of the vulcanizates at low temperatures, as compared to the sodium-butadiene rubber produced by the emulsion method. The properties of the acryl polymer were considerably improved at low temperatures upon introducing 1% of phosphorous into the polymer. The following conclusions could be drawn: the ethers of the unsaturated phosphene acids copolymer-Card 1/2

Copolymerization of

S/138/62/000/004/001/008 A051/A126

ize with the butadiene and butylacrylate, forming rubber-like and liquid poly-mers. The introduction of the phosphorous atom into the polymer chains of the sodium-butadiene and butylacrylate rubbers improves their properties at low temperatures, increases the resistance to various solvents and, in some cases, increases the physico-mechanical indices of the rubbers. There are 3 tables. The reference to the most recent English-language publication reads as follows: 3.M.C.Cormack, Pat. USA 2671078, 2671079, C.A., 48, 6738 (1954).

ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo kauchuka (All-union Scientific-Research Institute of Synthetic Rubber)

Card 2/2

KOROTKINA, D.Sh.; VINOGRADOVA, V.V.; KARELINA, G.G.

Copolymerization of unsaturated organophosphorus compounds. Kauch.i rez. 21 no.4:1-3 Ap '62. (MIRA 15:4)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo kauchuka.

(Phosphorus organic compounds) (Polymerization)

L 10748-63 EPR/EMP(j)/EPF(c)/EWT(m)/BDS--AFFTC/ASD--Ps-4/Pc-4/Pr-4--RM/WW/MAY

ACCESSION NR: AP3003286.

\$/0138/63/000/006/0006/0010

AUTHOR: Borisov, S. N.; Karelina, G. G.

70

TITLE: Dependence of vinylsiloxane rubber vulcanizates on the content of vinyl groups in elastomers

SOURCE: Kauchuk i rezina, no. 6, 1963, 6-10

APPROVED FOR RELEASE: 06/13/2000

TOPIC TAGS: polymerization, vulcanization, properties of vulcanizates, vinyl-phenylsiloxane rubbers, methylvinylsiloxane rubbers, tensile strength, elongation, heat resistance, low-temperature resistance, network density, deformation, MV-1,

ABSTRACT: The effect of the vinyl-group content on the properties of vinylsiloxane vulcanizates has been studied in detail. Dimethylsiloxane rubbers containing 0.5 to 10 mol% RC_2H_3SiO (where $R=CH_3$ or C_6H_5) groups were used. The monomers were synthesized by hydrolysis of equimolar mixtures of dimethyl- with methylvinylor vinylphenyldichlorosilane in calculated amounts of cyclodimethylsiloxanes and polymerized in the presence of concentrated sulfuric acid or aluminum sulfate dihydrate. Mixes containing 100 parts rubber and 50 parts U-333 light-colored filler

Card 1/82

CIA-RDP86-00513R000720710008-6"

L 10748-63

ACCESSION NR: AP3003286

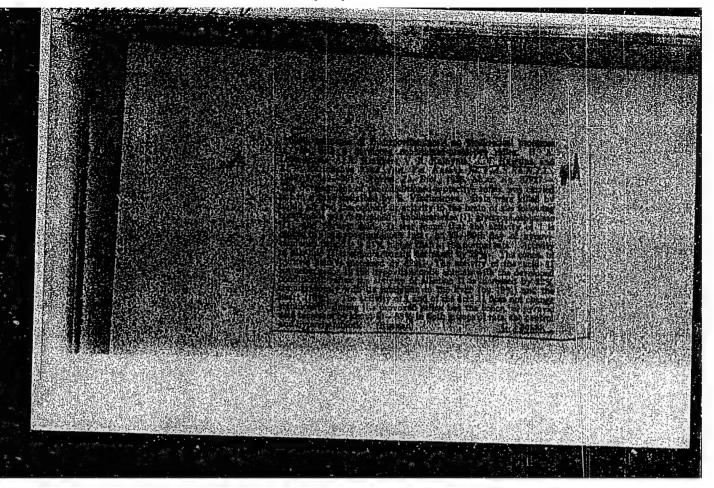
were vulcanized with benzoyl peroxide or sulfur and thiuram. The optimum dose of benzoyl peroxide for vinylpherylsiloxane rubbers (I) was found to be 0.4% regardless of the content of the vinylphenyl group. For effective sulfur vulcanization bthe vinyl-phenyl-group content in rubbers must be at least 1%. The optimum degree of vulcanization is attained faster for methylvinylsiloxane rubbers (II) than for I, in which the vinyl groups are hindered by phenyl radicals at the same Si atoms of the polymer chain. The tensile strength and elongation of sulfur and peroxide vulcanizates II and I containing 1 to 10% vinyl groups vary from 42 to 28 kg/cm2 and 255 to 140%. The heat resistance of I vulcanizates is somewhat higher than that of II vulcanizates, owing to the hindering effect of the phenyl groups; peroxide vulcanizates exhibit higher heat resistance than sulfur valcanizates. Vulcanizates of rubbers containing 1% vinyl groups (MV-1 and VF-1 rubbers) exhibit satisfactory properties after aging for 10 days at 2500. The low-temperature resistance of the vulcanizates is determined mainly by the second radical at the Si atom. It is higher in the presence of phenyl groups, which retard rubber crystallization. The low-temperature resistance at -600 of sulfur vulcanizates of MV-1 and VF-1 rubbers is higher than that of peroxide vulcanizates. Increasing the network density of MV-1 vulcanizates by using larger amounts of benzoyl peroxide increases their low-temperature resistance. Study of the deformation of vulcanizates after compression showed that it is lowest for peroxide vulcanizates of All Scientific Research of Synthetic Rubber.

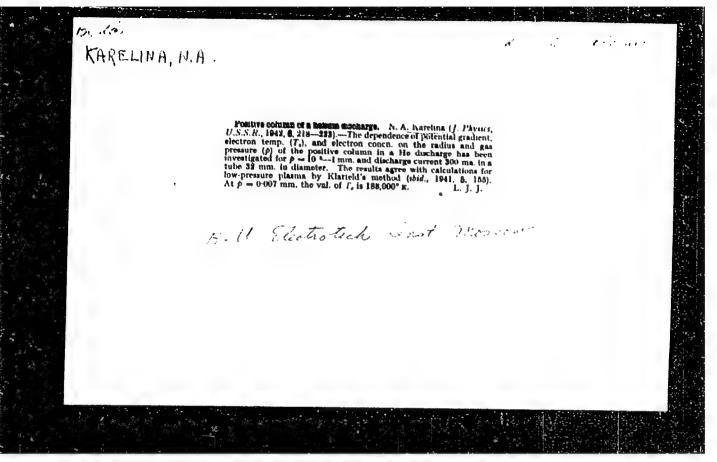
ROMASHKEVICH, I.F.; KARELINA, G.N.

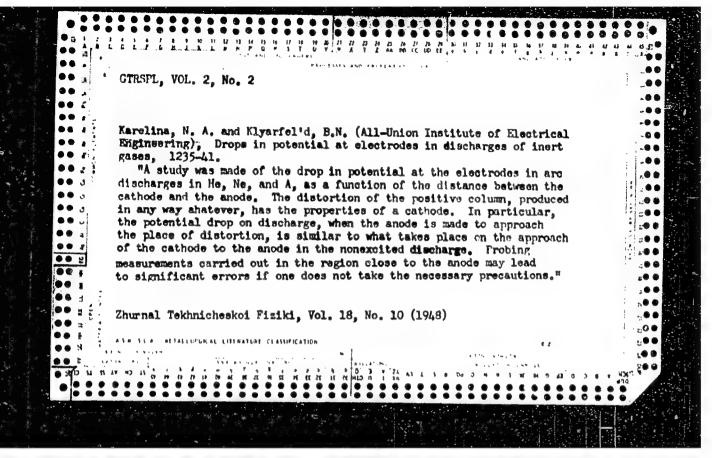
Production of methane and organic fertilizers through fermentation of wood waste and manure. Mikrobiologiia 30 no.1:146-151 Ja-F '61.

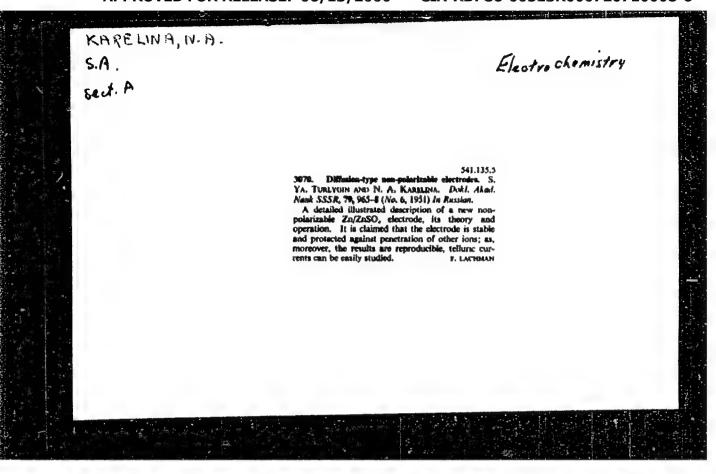
(MIRA 14:5) 1. Vsesoyuznyy naucino zpo-agropochvovedeniya, Moskva. (METHANE) (FERTILIZERS AND MANURES) 1. Vsesoyuznyy nauchno-issledovatel skiy institut udobreniy i

"APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000720710008-6









TURINGTU, S. WA.; MARRITHA, U.A.

Georhysics

Influence of dry land and sea on the distribution of natural electric currents in Earth's crust, Izv. AN SSSR. Ser. geofiz., No. 4, 1952.

Monthly Hast of Russian Accessions, Library of Congress, November 1952. UNCLASSIFIED.

Contact electrodes for measuring electrical current and voltage in the ocean. Trudy MOI 7:3-14 '56.

(Ocean) (Electric measurements)

TURLYGIN, S.Ya. [deceased]; KARPLINA, N.A.

Nonpolarizing diffusion electrodes for measuring weak currents occuring in any medium. Trudy MGI 7:15-26 '56. (MLRA 9:9) (Electric measurements) (Terrestrial electricity)

9.3120 (1003, 1137,1140)

21438

S/109/61/000/001/017/023 E140/E163

AUTHORS:

Yasnopol'skiy, N.L., Karelina, N.A., and Malysheva, V.S.

TITLE :

Certain results of the investigation of secondary electron emission from the backs of magnesium oxide

emitters

PERIODICAL: Radiotekhnika i elektronika, Vol.6, No.1, 1961,

pp. 146-152

TEXT: A thin-film secondary electron emitter permitting emission from the face opposite that irradiated by the primary electrons is described. Aluminium films between 100 and 1000 Å and MgO emitters deposited on them have been studied. It is shown that reduction of the Al thickness from 3000 to 350 Å permits reduction of the working potential from 11 - 18 to 3 - 4 kV at secondary emission factors of the order of 5 - 8. It has been found that under certain conditions the secondary emission from such targets can pass into a self-maintained emission. The device used is shown in Fig.1 in which K is the cathode, A - anode, $K_1 \sim$ collector of reflected primary electrons and forward emitted secondary electrons, β - secondary electron emitter, Card 1/3

7

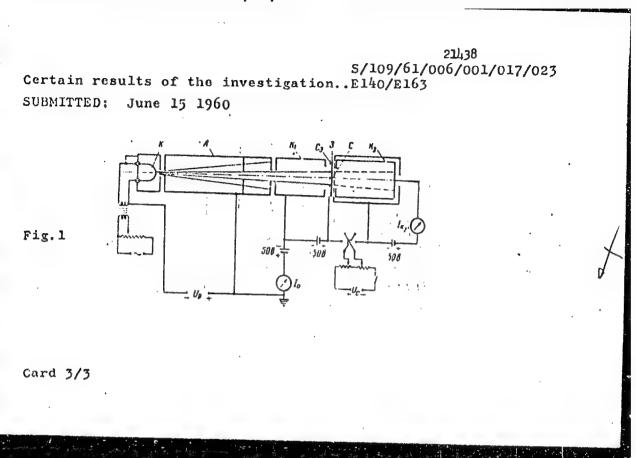
Card 2/3

21/138

\$/109/61/006/001/017/023 E140/E163

Certain results of the investigation of secondary electron emission from the backs of magnesium oxide emitters

Co - mesh base of emitter, C - grid for acceleration of electrons emitted from the back of the target, K_2 - collector for primary electrons passing completely through the emitter and secondary electrons emitted from the back. With certain potentials in this system it is found that the secondary emission from the back of the target will increase to a value of the order of 800 In and continue to flow after the primary beam is cut off. Initally this emission is relatively stable and easily excited. this behaviour deteriorates, apparently connected with impoverishment of the secondary-emission properties of the MgO layer. Possible explanations are connected with the formation of an autonomous discharge in a solid dielectric (Ref.6), and avalanche (Ref.7) or tunnel (Ref.8) emission under the influence of ion bombardment. A.I. Pyainitskiy, Ye.A. Krasovskiy, V.G. Butkevich and M.M. Butslov are mentioned for their contributions in this field. There are 8 figures and 8 references: 5 Soviet and 3 English.



AKOPOV, K.A.; KARELINA, N.A.; POKALYAKIN, V.I.; STEPANOV, G.V.

Interagency seminar on cathode electronics. Radiotekh.i
elektron. 6 no.5:863-864 My '61. (MIRA 14:4)
(Electronics—Congresses)

'n,

ACCESSION NR: AP4038615

8/0109/64/009/004/0643/0648

AUTHOR: Yasnopol'skiy, N.L.; Karelina, N.A.

TITIE: Effective secondary-electron emitter made of cesium-treated magnesium oxide and operating with shot-through primaries

SOURCE: Radiotekhnika ielektronika, v. 9, 1964, 643-648

TOPIC TAGS: secondary emission; secondary emission layer, cesium vapor treatment, magnesium oxide emitter

ABSTRACT: The response of compacted and uncompacted (friable) MgO emitters to treatment in cesium vapor was investigated with the aim of increasing the secondary emission coefficient. The emitters were to operate at low voltage with the primary electrons shot through the emitter. The technology of emitter preparation and the measurement procedure are described by the authors elsewhere (with V. S. Maly*sheva, Radiotekhnika i elektronika, 1961, v. 6, no. 1, 146). The tests resulted in a low-voltage effective emitter made of compacted layers of magnesium oxide treated in cesium vapor (see Fig. 1 of Enclosure). The secondary emission coefficient at 3 kev, for shot-through primaries, is about

Cord 1/4

ACCESSION NR: AP4038615

10, which is nearly double the coefficient of an untreated layer. In the case of uncompacted layers, cesium-vapor treatment contributes to the development of secondary emission, which is intensified by the field and can become self-maintaining emission (see Fig. 2 of Enclosure). "The authors are grateful to D.V. Zernov for a discussion of the work, and also to A.P. Ryabova for preparation of the experimental samples and instruments." Orig. art. has: 4!figures.

ASSOCIATION: none

SUBMITTED: 14Feb63

ENCL: 02

SUB CODE: EC

NO REF SOV: 006

OTHER: 004

Card 2/4

